

7 Science Coursework Planner

Term 4: Biological Sciences



Criterion Assessed: Understanding and Skills

Use of this coursework plan:

Use this coursework plan to inform your learning. You should *tick off* a topic as you learn and understand it and study it at home. Weekly homework is expected with well written sentences. Read **topic 3.2 and Ex 3.2 Q 1 – 4** means students need to read this section and complete the numbered questions for homework. Your answers should be of a higher standard than the simple answers provided by the textbook. These answers will be provided at the end of each week electronically.

© will be used for high level (A or B standard) concepts only. These items may not be addressed by all classes.

Summative Assessment: 1 x 70 min exam

WEEK	Elaborations of Content Descriptors Knowledge, concepts, skills and processes that students are expected to learn. Students will:	Guidance Assessment X3 Feedback x 3 Weekly Homework
1	Classification □Understand classification is based on physical characteristics and group organisms based on physical characteristics. □Understand classification knowledge has changed over time. □Classify organisms using hierarchical systems such as kingdom, phylum, class, order, family, genus and species. Website to check https://clickv.ie/w/UYuo	Coursework planner handed out Bookwork expectations delivered <u>Homework:</u> Read topic 3.2 and 3.4 Complete Ex 3.2 Q 1 – 4, 10 Something extra for class or home Complete investigation 3.2 'Living, non-living or dead?'
2	Dichotomous keys Use and apply scientific conventions for naming species. Organise and group a variety of organisms based on the basis of similarities and differences in particular features. Understand and apply the conventions of dichotomous keys. Examine, construct and use dichotomous keys. Identify and classify organisms using branching and written dichotomous keys. Website to check https://biologydictionary.net/dichotomous-key/	Homework: Read Topic 3.3 – 3.4 Complete Ex 3.4 Q 3 – 7 Something extra for class or home Complete investigation 3.4 'Making a class key'
3	Dichotomous keys □ Consolidate and extend understanding of classification ising dichotomous keys. □ © Use a learning object to classify plants or bacteria. Websites to check https://youtu.be/wpKulkADzBk https://fergusonfoundation.org/resources/game-fishing-for-a-name/	Class quizzes / warm-ups with feedback <u>Homework:</u> Read Topic 3.3 – 3.6 Complete Ex 3.5 Q 1 – 6 © 22a.
4	Food chains □ Construct and use food chains to represent feeding relationships in a habitat. □ Classify organisms in an environment according to their position in a food chain. □ Recognise the role of microorganisms within food chains. Websites to check https://www.nationalgeographic.org/topics/resource-library-food-chains-and-webs/?q=&page=1&per_page=25 https://clicky.ie/w/-cvo	Bookwork check <u>Homework:</u> Read Topic 4.1 – 4.4 Complete Ex 4.4 Q 1 – 5 Something extra for class or home Complete Worksheet 4.5 'Food chains and food webs' on Readcloud

5	Food webs	Formative assessment
	Identify an organism's trophic level.	Ongoing feedback
	□ Understand that the number of organisms in each trophic level decreases as the	
	level increases.	Homework:
	□ <i>Recognise</i> the role of microorganisms within food webs.	Read Topic 4.1 – 4.4
	□ Interpret food webs to show relationships between organisms in an environment.	$Complete Ex 4.4 Q 7, 8 \oplus 14$
	□ Construct, use, and interpret food webs to show relationships between organisms	
	in an environment.	Something extra for class or home.
		Complete a food web using
	Websites to check	Worksheet 4.2 'Food webs' on
	https://youtu.be/hLq2datPo5M	Readcloud
	https://www.bbc.co.uk/bitesize/guides/zq4wjxs/revision/1	
6	Examining human impact	Class quizzes / warm-ups with
	□ <i>Identify</i> key species within a habitat.	feedback
	□ <i>Identify</i> trophic levels in food pyramids.	Homowork
	□ Understand human activity has a range of impacts on a food web.	Read Topic 4 4 – 4 9
	□ <i>Analyse</i> and <i>interpret</i> data showing population numbers.	Complete Fx 4.4 0 8 © 16
	\Box \odot <i>Elaborate</i> on the effects of human activity beyond food webs.	Complete Ex 4.9 Q 24 – 26
	Websites to check	
	https://www.pbslearningmedia.org/resource/human-impact-food-webs-video-	
	gallery/the-age-of-nature/	
7	Investigating and evaluating human impact	
	□ <i>Examine</i> and <i>explore</i> the impact of introduced species on other living things.	Homework:
	\Box Investigate the effect of human activity on local habitats, such as deforestation,	
	agriculture or the introduction of new species.	Read Topic $4.9 - 4.12$
	□ Understand the factors that cause species to become 'endangered'.	
	□ Investigate needs and means for controlling introduced species.	
	□ <i>Identify</i> the different levels of 'threatened' species.	Something extra for class or home
	\Box \bigcirc Discuss the benefits and arguments surrounding human engagement with	Complete Worksheet 4.7 'Population
	'threatened' species e.g. zoos, scientific testing, cultural practices.	overload' on Readcloud
	\Box $\textcircled{\odot}$ Research specific examples of human activity, such as the effects of palm oil	
	production.	Complete Worksheet 4.8 'Spot the
		pest' on Readcloud
	Websites to check	
	https://www.science.org.au/curious/earth-environment/invasive-species	
0	https://www.nationalgeographic.org/encyclopedia/endangered-species/	
0	Native food webs	Homework:
		<u>Homework.</u>
	Conduct the 'Maccuring abiatic factors' prac	Read Topic 4.7
	Conduct the Measuring ablotic factors prac.	Complete Ex 4.7 Q 2, 3, 4a – d 😊
	factors	
	Websites to check	
	https://www.natureaustralia.org.au/what-we-do/our-	
	insights/perspectives/human-impact-nature-australia/	
	https://clickv.ie/w/2dvo	
	https://www.youtube.com/watch?v=q2zdiLn3gSE	
9	Revision	Summative assessment
	□ Consolidate and extend understanding of classification	Exam
	□ <i>Review</i> the conventions for constructing branching and written dichotomous keys	Exam feedback
	Assessment	
	Exam (Understanding and skills)	
10	Deview and reflect	
10	Review and reflect	
	Achievement in Year 7 Science discussed and goals set for Year 8 and hevend	
	Achievement in fear 7 science discussed and goals set for fear 8 and beyond	

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SCIENCE DEPARTMENT – YEAR 8 COURSEWORK PLANNER

TOPIC FOUR – Biological Sciences

Building Blocks of Life and Reproduction (10 weeks)

<u>Summative Assessment:</u> 1 x 70 min Written Exam <u>Criteria Assessed:</u> Understanding and Skills

WEEK	Elaborations of Content Descriptors Knowledge, concepts, skills and processes that students are expected to learn. Students will :	Guidance Assessment x3 Feedback x3 Weekly Homework
1	Unicellular Life & Exploring the microscope	Quiz Prior Knowledge
	 Be able to define the term cell. Be able to distinguish between unicellular and multicellular life forms. 	Homework
	 Identify parts of a microscope (stage, light, objective lenses, ocular lens, base. 	Read 3.1 to 3.3
	fine focus knob, coarse focus knob, and clips) and be able to explain their	Complete Ex 3.2 Q1,2,4
	purpose.	EX 3.3 Q1, 11, 12
	Understand magnification.	
	 Be able to calculate the real size of an object given its apparent size and magnification 	
	Videos	
	https://online.clickview.com.au/exchange/videos/26511383/microscopes-	
	and-cells	
	nttps://online.clickview.com.au/libraries/videos/3716289/using-a-	
2	Multicellular life : Parts of an animal cell	
2	Be able to prepare a wet mount slide and safely observe slides under a	
	microscope.	Homework
	Be able to draw a scientific diagram of objects under a microscope.	Complete Ex 3.4 O1 to 5, 12
	Use a microscope to observe, identify and illustrate pond life.	
	Onderstand the levels of organisation: cells, tissues, organs, system. Be able to identify general plant cell organelles: cell membrane, nucleus	
	mitochondria. cvtoplasm. cell wall. chloroplasts. and vacuole and describe	
	their function.	
	• © Explore examples of specialised plant cells (e.g. xylem, phloem, mesophyll	
	cell and guard cells), their structure and function.	
	https://online.clickview.com.au/exchange/categories/6364/biology/videos/58468	
	17/cell-organelles	
3	Animal Cells	Class Quizzes/ feedback
5	• Describe specialised animal cells (e.g. muscle, nerve and blood cells), their	Homework
	structure and function.	Read Topic 3.6
	• Be able to identify animal cell organelles. cell membrane, cytopiasm, nucleus, mitochondria (© ribosomes endonlasmic reticulum)	Complete Ex 3.6 Q1,2,4 & 5
	 Describe the function of the above animal cell organelles. 	
	 O Make a 3-D representation of a cell (using jelly for example). 	
	Be able to link the function of specialised cells with the variety of their	
	organelles, e.g. mitochondria and muscle cells.	
	 Distinguish between bacteria, animal and plant cells and be able to compare prokaryotes and eukaryotes 	
	 Be able to compare and contrast plant and animal cells. 	
	Videos	
	https://online.clickview.com.au/libraries/videos/23485886/animal-and-	
	Plant Reproduction	Class Quizzes/feedback
4	Dissect and be able to identify parts of a flower such as netal filament	
	stigma, style, anther, sepal, pollen, ovary, ovule, stamen, and carpel.	Homework
	Know the function of each part of a flower.	Read Topic 3.7, 3.8, 5.11
	• Understand the difference between types of pollination and be able to identify	Ex. 3.8 Q1.5.7
	which system a plant uses. E.g. self and cross pollination, animal and wind	Ex 5.11 Q1,3,4
	Videos	
	https://online.clickview.com.au/libraries/videos/15495086/the-structure-of-	
	typical-flowers	
	https://online.clickview.com.au/libraries/series/15426177/reproduction-in-	
	plants/videos/3/16/46/tertilisation-in-flowering-plants https://online.clickview.com.au/libraries/series/15426177/reproduction-in-	
	plants/videos/3716731/pollination	
1		

5		Human Reproduction	Homework
5	٠	Be able to identify the following male and female anatomical parts and describe	Read Topic 5.2
		their function: vagina, ovary, fallopian tube, uterus, cervix, clitoris, urethra,	Complete EX 5.2 Q4,5
		ureter, kidney, bladder, penis, testis, epididymis, vas deferens, prostate	
		gland.	
6		Human Reproduction	Class Quizzes /feedback
Ŭ	•	Know the physical changes that puberty causes in males and females.	Homework Road Tapia 5.2
	•	Know that changes in puberty are due to hormones.	Read Topic 5.5 Complete Ex 5.2 \bigcirc 2.8 \checkmark
	•	③ Be able to describe the functions of FSH, oestrogen and progesterone.	Complete EX 5.5 QZ,5 & 4
		Videos	
		https://online.clickview.com.au/libraries/categories/3708551/videos/371450	
-		<u>3/human-reproduction-and-childbirth</u>	
7		Comparing plant & animal Reproductive cells	Mix n Match/ Consolidation ppts
'	•	Be able to describe the menstrual cycle and identify the timing of ovulation.	
	•	\odot Be able to link the levels of hormones and thickness of endometrium to	Read Topic 5.6 & 5.7 Complete Ex 5.7 01.2
		stages of the menstrual cycle through the interpretation of diagrams.	$E_{x} 5 8 \cap 12 3$
	•	Compare the function of reproductive cells in plants and animals - sperm, eggs,	EX 3.0 Q1,2,3
		pollen, and ovum.	
	•	Image: University of the second se	
	•	© Contraception	
	٠	© Pregnancy Complications (ectopic pregnancy etc.)	
0		Revision & Assessment	Exam/ Feedback
0	٠	Exam testing Science Understanding and Skills.	
		Immunity	
9/10	•	Describe the various types of defence against infection such as physical	
	-	barriers, inflammation and white blood cells	
	•	© Explain the roles of the lymphatic system. T lymphocytes, antibodies and	
		antigens.	
	•	Be able to define pathogens and hosts.	
		Explore disease causing organisms such as bacteria, viruses, protozoa	
		Videos	
		https://online.clickview.com.au/exchange/categories/13000/health/videos/2	
		854333/the-immune-system	
		https://online.clickview.com.au/exchange/categories/6364/biology/videos/	
		<u>38288/parasites</u>	

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SCIENCE DEPARTMENT – YEAR 9 COURSEWORK PLANNER

TOPIC FOUR – Chemical Sciences

Chemical Patterns and Heat and Eat units (10 weeks)

Use this coursework plan to inform your learning. You should tick off a topic as you learn and understand it and study it at home. Weekly homework is expected with well written sentences. **Topic 3.4 and Ex 3.4 Q 1-3, 5** means students need to read this section and complete the numbered questions for homework. Your answers should be of a higher standard than the simple answers provided by the textbook. These answers will be provided at the end of each week electronically. We have included some interesting optional websites to assist you. ⁽²⁾ will be used for high level (A or B standard) concepts only. These items may not be addressed by all classes.

WEEK	Elaborations of Content Descriptors Knowledge, concepts, skills and processes that students are expected to learn. Students will :	Assessment x 3 Feedback x 3 Weekly Homework
1	Chemical Reactions and Conservation of Mass Revise the names and symbols of the first twenty elements. Know that in a chemical reaction, reactants are transformed into products. Know that an exothermic reaction or process gives off energy (usually as heat), while an endothermic reaction or process absorbs energy. Know the difference between a chemical reaction and a physical process. Know the Law of Conservation of Mass – in a chemical reaction, the total mass of the reactants is the same as the total mass of the products. Know that in a chemical formula the subscripts tell you the number of atoms, e.g. CuSO₄ has one copper atom, one sulphur atom and four oxygen atoms. Se able to <i>solve</i> and balance chemical equations. Websites to check: https://www.khanacademy.org/science/ap-chemistry/stoichiometry-and-molecular-composition-ap/balancing-chemical-equations-ap/v/chemical-reactions-introduction https://www.youtube.com/watch?v=3lHHOiTdmK4	Coursework planner handed out. Bookwork expectations delivered. Topic 7: Chemical Reactions (page 319) Homework: 1) Read the coursework planner 2) Cross-Reference the coursework planner and Readcloud/ <u>new version</u> of the textbook. Read Topic: 7.1 and 7.2 (pages 319, 321 - 322) Complete Ex: 7.2, Q1, 2, 3, © 6, © 8 (page 323) Read Topic: 7.3 (pages 324-326) Complete Ex: 7.3, Q1, © 8 (page 327)
2	Reactions of acids and bases List the properties of acids: taste sour, react with active metals, and that some acids are corrosive. Know some common substances which contain acids, e.g. vinegar, lemon juice, ant bites, yoghurt. List the properties of bases: taste bitter, feel slippery or soapy, and that some bases are corrosive. Know some common substances which contain bases, e.g. drain cleaner, dishwashing liquid. Observe that acids react with active metals to make hydrogen gas and a salt. active metal + acid → H₂ + salt ③ Know that acids are substances that dissociate to release hydrogen ions in water, and bases release hydroxide ions in water.	Homework: Read Topic: 7.4 (page 328) Complete Ex: 7.4, Q1, 2, 3, 12 (page 333, 334)
	Websites to check: <u>https://www.youtube.com/watch?v=3IHHOiTdmK4</u>	

	https://www.khapacademy.org/science/chemistry/acids-and-bases-	
	topio	
	<u>topic</u>	
	Acids and Bases	Assessment handed out
		Assessment nanded out
	Know that a neutral substance is neither acidic nor basic.	A-level Exemplar handed
	Know that an alkali is a base dissolved in water.	out
	Know that an indicator is a substance that is one colour in a very acidic	
	where and a different solaur is a year basis solution. Different indicators	
	solution and a different colour in a very basic solution. Different indicators	Homework:
3	change colour at different pH levels.	
5	Compare the acidities of solutions by using the pH scale; which is a scale	©begin assignment by
	from 1 (very acidic) to 14 (very basic / alkaline).	researching theory at home
		3 1 1 1
	Websites to shock:	Read Tonic: 7 4 (nage 228)
	Websites to check.	Neau Topic. 7.4 (page 520)
		Complete Ex: 7.4 Q4, 8, 9,
	https://phet.colorado.edu/en/simulation/ph-scale-basics	10. © 13. © 15 (pg 333, 334)
	Assessment	Homework:
		Homework.
		©continue assignment by
4	Conduct and write up Assessment Task	researching theory and
	Image: Be able to interpret Safety Data Sheets (SDS)	writing report at home
		writing report at nome
	Acids and Bases	
	Know that acids and bases neutralise each other.	Homework:
	Know the word equation for neutralisation :	Read Topic: 7.4 (page 328)
	acid + base \rightarrow water + a salt	
5		
	Determine the nH of substances by using an indicator	Complete Ex: 7.4 05. 6.
	Determine the pH of substances by using an indicator.	Complete Ex: 7.4 Q5, 6, @ 14_@ 18 (nage 333_334)
	Determine the pH of substances by using an indicator.	Complete Ex: 7.4 Q5, 6, © 14, © 18 (page 333, 334)
	Websites to check:	Complete Ex: 7.4 Q5, 6, © 14, © 18 (page 333, 334)
	Websites to check:	Complete Ex: 7.4 Q5, 6, © 14, © 18 (page 333, 334)
	Determine the pH of substances by using an indicator. Websites to check: <u>https://phet.colorado.edu/en/simulation/acid-base-solutions</u>	Complete Ex: 7.4 Q5, 6, © 14, © 18 (page 333, 334)
	Determine the pH of substances by using an indicator. Websites to check: <u>https://phet.colorado.edu/en/simulation/acid-base-solutions</u>	Complete Ex: 7.4 Q5, 6, © 14, © 18 (page 333, 334)
	Determine the pH of substances by using an indicator. Websites to check: <u>https://phet.colorado.edu/en/simulation/acid-base-solutions</u>	Complete Ex: 7.4 Q5, 6, © 14, © 18 (page 333, 334)
	Determine the pH of substances by using an indicator. Websites to check: <u>https://phet.colorado.edu/en/simulation/acid-base-solutions</u>	Complete Ex: 7.4 Q5, 6, © 14, © 18 (page 333, 334) DRAFT DUE
	Determine the pH of substances by using an indicator. Websites to check: https://phet.colorado.edu/en/simulation/acid-base-solutions Acids and Bases	Complete Ex: 7.4 Q5, 6, © 14, © 18 (page 333, 334) DRAFT DUE Homework:
	Determine the pH of substances by using an indicator. Websites to check: https://phet.colorado.edu/en/simulation/acid-base-solutions Acids and Bases © Explain that acid rain is caused when sulphur-containing substances are	Complete Ex: 7.4 Q5, 6, © 14, © 18 (page 333, 334) <u>DRAFT DUE</u> Homework:
	Determine the pH of substances by using an indicator. Websites to check: https://phet.colorado.edu/en/simulation/acid-base-solutions Acids and Bases © Explain that acid rain is caused when sulphur-containing substances are burnt, making sulphur diavide which reacts with water in the atmosphere	Complete Ex: 7.4 Q5, 6, © 14, © 18 (page 333, 334) <u>DRAFT DUE</u> Homework: <i>© continue assignment by</i>
	Determine the pH of substances by using an indicator. Websites to check: https://phet.colorado.edu/en/simulation/acid-base-solutions Acids and Bases © Explain that acid rain is caused when sulphur-containing substances are burnt, making sulphur dioxide which reacts with water in the atmosphere.	Complete Ex: 7.4 Q5, 6, © 14, © 18 (page 333, 334) <u>DRAFT DUE</u> Homework: © continue assignment by researching theory and
	Determine the pH of substances by using an indicator. Websites to check: https://phet.colorado.edu/en/simulation/acid-base-solutions Acids and Bases © Explain that acid rain is caused when sulphur-containing substances are burnt, making sulphur dioxide which reacts with water in the atmosphere. © Know that strong acids or bases are substances that dissociate completely	Complete Ex: 7.4 Q5, 6, © 14, © 18 (page 333, 334) <u>DRAFT DUE</u> Homework: © continue assignment by researching theory and start writing report at
	Determine the pH of substances by using an indicator. Websites to check: https://phet.colorado.edu/en/simulation/acid-base-solutions Acids and Bases © Explain that acid rain is caused when sulphur-containing substances are burnt, making sulphur dioxide which reacts with water in the atmosphere. © Know that strong acids or bases are substances that dissociate completely in water, and that weak acids and bases do not dissociate completely in	Complete Ex: 7.4 Q5, 6, © 14, © 18 (page 333, 334) <u>DRAFT DUE</u> Homework: © continue assignment by researching theory and start writing report at home
6	Determine the pH of substances by using an indicator. Websites to check: https://phet.colorado.edu/en/simulation/acid-base-solutions Acids and Bases © Explain that acid rain is caused when sulphur-containing substances are burnt, making sulphur dioxide which reacts with water in the atmosphere. © Know that strong acids or bases are substances that dissociate completely in water, and that weak acids and bases do not dissociate completely in water. Some weak acids are weaker than others, because they dissociate	Complete Ex: 7.4 Q5, 6, © 14, © 18 (page 333, 334) <u>DRAFT DUE</u> Homework: © continue assignment by researching theory and start writing report at home
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6	Determine the pH of substances by using an indicator. Websites to check: https://phet.colorado.edu/en/simulation/acid-base-solutions Acids and Bases © Explain that acid rain is caused when sulphur-containing substances are burnt, making sulphur dioxide which reacts with water in the atmosphere. © Know that strong acids or bases are substances that dissociate completely in water, and that weak acids and bases do not dissociate completely in water. Some weak acids are weaker than others, because they dissociate less. © Justify that a dilute solution is a solution containing not much solute in	Complete Ex: 7.4 Q5, 6, © 14, © 18 (page 333, 334) <u>DRAFT DUE</u> Homework: <i>© continue assignment by</i> <i>researching theory and</i> <i>start writing report at</i> <i>home</i> Read Topic: 7.5
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6	Determine the pH of substances by using an indicator. Websites to check: https://phet.colorado.edu/en/simulation/acid-base-solutions Acids and Bases © Explain that acid rain is caused when sulphur-containing substances are burnt, making sulphur dioxide which reacts with water in the atmosphere. © Know that strong acids or bases are substances that dissociate completely in water, and that weak acids and bases do not dissociate completely in water. Some weak acids are weaker than others, because they dissociate less. © Justify that a dilute solution is a solution containing not much solute in lots of water, whereas a concentrated solution contains lots of solute in not much water. © Be able to use these concepts to compare solutions e.g. a dilute solution	Complete Ex: 7.4 Q5, 6, © 14, © 18 (page 333, 334) <u>DRAFT DUE</u> Homework: <i>© continue assignment by</i> <i>researching theory and</i> <i>start writing report at</i> <i>home</i> Read Topic: 7.5 (pages 335 - 336) Complete Ex: 7.5 Q1, 3, © 5 (page 337)
6	Determine the pH of substances by using an indicator. Websites to check: https://phet.colorado.edu/en/simulation/acid-base-solutions Acids and Bases © Explain that acid rain is caused when sulphur-containing substances are burnt, making sulphur dioxide which reacts with water in the atmosphere. © Know that strong acids or bases are substances that dissociate completely in water, and that weak acids and bases do not dissociate completely in water. Some weak acids are weaker than others, because they dissociate less. © Justify that a dilute solution is a solution containing not much solute in lots of water, whereas a concentrated solution contains lots of solute in not much water. © Be able to use these concepts to compare solutions e.g. a dilute solution of a strong acid to a concentrated solution of a weak acid.	Complete Ex: 7.4 Q5, 6, © 14, © 18 (page 333, 334) <u>DRAFT DUE</u> Homework: <i>© continue assignment by</i> <i>researching theory and</i> <i>start writing report at</i> <i>home</i> Read Topic: 7.5 (pages 335 - 336) Complete Ex: 7.5 Q1, 3, © 5 (page 337)

7	Reactions of metals ③ Be able to solve and balance complex chemical equations. Know that acids react with active metals to make hydrogen gas and a salt. Know that acids react with active metals to make hydrogen gas and a salt. Know that the more reactive the metal, the faster that reaction will go. Know that an activity series lists metals in order of reactivity. Know that in a displacement reaction, the more reactive metal ends up in solution. Websites to check: https://phet.colorado.edu/en/simulation/balancing-chemical-equations	Homework:
8	 Reactions of metals and Food chemistry [©] Be able to <i>use</i> an activity series of metals to predict the results of a displacement reaction. [©] Be able to use the results of a set of displacement reactions to categorise the metals in order from most reactive to least reactive. Know that metals can be extracted from their oxides by smelting. [©] Analyse how the physical properties of metals can be changed by quenching, annealing and tempering. Know that acids react with carbonates to make carbon dioxide gas, water and a salt; and that this is used in many applications including baking. Conduct experiment on reactions involving antacids 	Submit completed assessment Homework: Read 7.9 (Review) P.343 Complete 7.9 Review Q1, 2, 5, 8, 12,
9 10	Combustion Chemistry Know that combustion is another name for burning. Know that in a combustion reaction, a substance reacts with oxygen to produce oxides. Determine how the products of combustion reactions affect the environment. Revision of chemical reactions and metals Higher Order Thinking	Homework: Read Topic: 7.6 (pages 338 – 339) Complete Ex: 7.6 Q1, 3, 4, © 6, © 12 (pages 339 – 340) Scientific report feedback
	Activities	